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The Role of Phosphate-Solubilizing Microorganisms in Soil Health and Phosphorus Cycle: A Review

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ABSTRACT: The role of phosphate-solubilizing microorganisms (PSM) in soil health and
Phosphorus (P) cycle is a crucial aspect of agricultural productivity. Recent research has emphasized
the importance of microorganisms in maintaining soil health. Phosphorus is an essential
macronutrient for plant growth and development and significantly affects crop productivity. The
PSM solubilize phosphate by producing metabolites such as organic acids, inorganic acids, hydrogen
sulphide,-exopolysaccharide, and siderophore. Studies have shown that the combination of PSM with
phosphate fertilizers increase the efficiency of the fertilizers in soils. The efforts have been made to
integrate exogenous soil microorganisms into P cycling models. This review highlights the critical
role of microorganisms in maintaining soil health and promoting P cycle, emphasizing the
importance of incorporating them into soil management practicesCorresponding Author:
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mechanism.

INTRODUCTION

Agriculture is fundamental to human beings and the constitution of society (Kavanagh et al., 2018). Soil health is essential for sustainable agricultural productivity including food production. Microorganisms play a critical role in maintaining soil health and promoting plant growth. Among these microorganisms, phosphate-solubilizing microorganism (PSM) are gaining attention due to their ability to solubilize phosphorus in the soil, making it more available to plants. As a result, PSM are natural alternatives for chemical fertilizers to improve soil health and P cycle in agricultural systems.

Phosphorus is an essential macronutrient for plant growth and development, and its availability in the soil significantly affects crop productivity. The use of high-concentration phosphate fertilizers continuously has become a common practice that threatens natural resources (Silva et al., 2023). Besides, most of the fertilizers applied to the soil become unavailable for assimilation by plants and can even lead to biological imbalances in soil and water (Blanco-Vargas et al., 2020; Silva et al., 2019). However, the majority of phosphorus in the soil is present in insoluble forms, making it unavailable to plants. The PSM can solubilize this insoluble phosphorus by secreting organic acids, phosphatases, and other metabolites, thereby making it available to plants.

The use of PSM in agriculture has several advantages, including improving plant growth, increasing nutrient uptake efficiency, and reducing the dependence on chemical fertilizers. Moreover, the use of PSM can reduce environmental contamination caused by excessive use of chemical fertilizers, which can lead to water bodies eutrophication; soil acidification, and hence soil degradation. These microorganisms also enable to solubilization of phosphate in acidic and alkaline soils and degrade xenobiotic compounds (Silva et al., 2023). PSM plays an important role in improving soil quality that associated with its ability to increase nutrient availability (Rafique et al., 2017) and reduce excessive use of inorganic fertilizers (Sharma et al., 2013). This study addresses the importance of phosphate solubilizing microorganism and its role in soil health and P cycle. In this way, this study presents bibliographic review about phosphate solubilization, metabolites produced by MPF, inorganic fertilizer efficiency, P cycle, and soil health.

Soil Phosphorus Cycle

In most soils, the concentration of total P is higher than that of other essential nutrients like nitrogen (N) and potassium (K). However, over 80% of P is immobile and not readily available to plants (Xu et al., 2020). Phosphorus is present in different

forms in the soil, primarily as inorganic P and organic P, and the distribution between these two forms varies as the soil matures (Figure 1). The significance role of the soil microorganism to influence soil P dynamics is recognized in agriculture. Therefore, exogenous microbial incorporation will maintain P cycling.

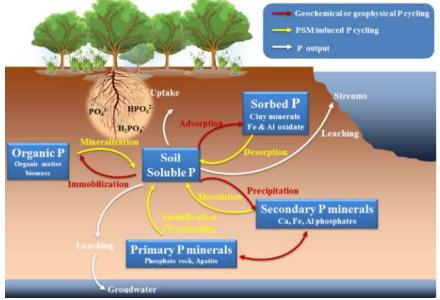


Figure 1. Soil phosphorus (P) cycle (Tian et al., 2021)

The atmosphere has limited involvement in P movement since P and its compounds are primarily in solid form at the standard temperature and pressure ranges found on Earth (Kumar et al., 2018). A significant portion of total soil P is contributed by organic P, which primarily originates from biological tissues where P is a constituent element of organic compounds such as nucleotides, phosphoproteins, phospholipids, and coenzymes. Compared to organic P, inorganic P in soil is present in a relatively insoluble and stable form. The P ions in unoccupied are attracted to the surfaces of Fe and Al oxides, and they can be easily leached. In contrast, occluded P is incorporated into layers and concretions of developing Fe and Al oxides during diffusive penetration and soil evolution, making it less easily leached (Tian et al., 2021).

Inorganic P is present in soil in a variety of forms and proportions (Mathew et al., 2020). It may leach into rivers to store P in marine sediments, or absorbed by plants or soil microbes to participate in the secondary organic phosphorus cycle. Long-term geochemical processes, such as weathering, adsorption/desorption, precipitation/dissolution, and solid-phase transformation, influence the form and distribution of P in soil in the majority of natural ecosystems (Hou et al., 2018).

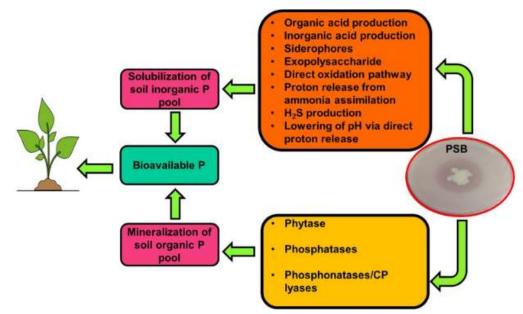
Soil organic originates from various biomolecules such as nucleotides, phosphides, co-enzymes, phosphoproteins, sugar phosphates, and phosphonates (Tamburini et al., 2012). These organic P compounds have relatively short lifespans and can make up to 65% of the total P content in typical soils (Dodd & Sharpley, 2015; Fabianska et al., 2019). Soil soluble orthophosphate ions can be immobilized in microbial cells to improve biomass growth (Tian et al., 2021). It is found that most of the P mineralized from organic P by PSM is incorporated into the bacterial cells as cellular P (Tao et al., 2008).

Controlling the release of orthophosphate from organic P sources in the soil is a significant aspect of the soil P cycle. Soil microbes, especially PSM, can enhance soil organic P cycle through organic P mineralization and decomposition (Tian et al., 2021). These biogeochemical processes are mainly moderated by the activities of phosphatase enzymes in PSM and soils (Sun et al., 2020).

The organic P hydrolysis activities of extracellular phosphatase enzymes are affected by soil properties, microbial interactions, plant cover, and environmental inhibitors and activators (Nannipieri et al., 2011). While soil PSM plays a role in organic P mineralization and the P cycle across different levels, the predominant enzymes and functional genes involved tend to be consistent (Tian et al., 2021). The relationship between microbial genes and the expression of phosphatase enzymes is recognized as the fundamental mechanism governing the conversion of organic P into bioavailable orthophosphates by PSMs (Bi et al., 2018).

Mechanism of Phosphate Solubilization by PSM and Metabolites Produced

The P availability in soil solution controlled by chemical processes (precipitation and dissolution of primary and secondary minerals), physico-chemical processes (adsorption and desorption of P from clays, oxides, and minerals) and biological processes (Frossard et al., 2000; Shen et al., 2011; Sims et al., 2002). The Biological mechanisms to control inorganic P availability in soil mineralization, solubilization, and immobilization are the predominant ways of dissemination of phosphorus in soil by PSMs, which are influenced by available inorganic minerals in the soil (Rawat et al., 2020).



Phosphate solubilization by PSM (Rawat et al., 2020)

PSM solubilize phosphate by producing some metabolites, including organic acids, inorganic acids, Hydrogen sulphide (H_2S) , exopolysaccharide (EPS), and siderophore. Organic acids released by PSM are citric acid, gluconic acid, oxalic acid, and tartaric acid (Table 1). These acids dissolve inorganic phosphates through several mechanisms (Kishore et al., 2015): (a) chelation of cations bound to phosphate, (b) reducing pH, (c) complexation with metal ions bound to phosphates, and (d) challenging P for adsorption site.

Microbes	Organic Acids	References
Pseudomonas spp.	Gluconic, formic, and citric acid	Pande et al., 2017
Pseudomonas prosekii	Gluconic, caprylic, fumaric,	Yu et al., 2019
	carboxylic, and benzoic acid	
Pseudomonas poae	Gluconic and malic acid	Vyas & Gulati, 2009
Pseudomonas fluorescens	Gluconic, lactic, dan succinic acid	Vyas & Gulati, 2009
Pseudomonas trivialis	Gluconic, lactic, succinic, formic,	Vyas & Gulati, 2009
	dan malic acid	
Bacillus sp.	Citric, malic, succinic, fumaric,	Selvi et al., 2017
	tartaric, and gluconic acid	
Proteus sp.	Citric, succinic, fumaric, and	Selvi et al., 2017
	gluconic acid	
Aspergillus sp.	Citric, gluconic, oxalic, succinic,	Sane & Mehta, 2015
	malic, and glycolic acid	
Azospirillum sp.	Citric, succinic, fumaric, and	Selvi et al., 2017
	gluconic acid	
Penicillium sp.	Gluconic, glycolic, succinic, malic,	Sane & Mehta, 2015
	oxalic, and citric acid	
Erwinia herbicola	Gluconic acid	Kumar et al., 2018
Bacillus megaterium	Gluconic, lactic, succinic, and	Saeid et al., 2018
	propionic acid	
Bacillus subtilis	Lactic, succinic, and propionic acid	Saeid et al., 2018
Enterobacter sp.	Acetic and malic acid	Suleman et al., 2018

Tabla 1	Organic acide	produced by	Phoenhoto	Solubilizing Microbes
Table 1.	Organic actus	produced by	Phosphate	Solubilizing Microbes

In addition to organic acids, PSM also produces inorganic acids. According to reports, PSMs generate nitric acid, carbonic acid, sulfuric acid, and hydrochloric acid that dissolve phosphate, but these acids are not as efficient as organic acids (Rawat et al., 2020). Nitrobacter and Thiobacillus produce inorganic acids such as nitric acids and sulfuric acid, respectively, to solubilize

phosphates (Shrivastava et al., 2018). Another mechanism to dissolve inorganic P involves the production of H_2S , which reacts with ferric phosphate to generate ferrous sulfate and release phosphate (Florentino et al., 2016).

Microorganisms produce EPS in harsh condition that lead to biofilm formation. The EPS constitute a substantial portion of bacterial mass within the extracellular matrix, typically accounting for anywhere between 40% and 95% (Flemming & Wingender, 2001). The presence of EPS contributes to an enhanced solubilization of phosphorus (P) by disrupting the P-solubilization homeostasis, leading to a greater release of phosphorus from insoluble phosphate compounds (Yi et al., 2008). Complex of EPS and metals is present in soil indicating their involvement in the solubilization of phosphorus in the soil (Ochoa-Loza et al., 2001).

Microorganisms also produce siderophores which are low molecular weight high-affinity iron chelating compounds that are excreted in response to iron stress in the environment (Rawat et al., 2020). The production of siderophores by PSB indirectly enhances the availability of phosphorus, as these ligands also have the capability to extract iron from ferric citrate and ferric phosphate (Zaidi et al., 2009). According to a report, in alkaline conditions, several phosphate solubilizers including *Rhizobium radiobacter*, *Pantoea allii*, *Bacillus subtilis*, and *Bacillus megaterium* produced siderophores ranging from 80 to 140 µmol/L which facilitate the survival of organisms in stressful environments and enhanced the solubilization of phosphorus (Ferreira et al., 2019).

Soil Health Improvement by PSM

Microorganisms have the ability to solubilize the insoluble phosphates and maintain the soil health and quality (Richardson, 2001). Rhizospheric microorganisms mediate soil process such as decomposition, nutrient mobilization and mineralization, and storage release of nutrients and water (Khan et al., 2007). Inoculation of *Bacillus cereus* and *Pseudomonas moraviensis*, exhibited 35% higher soil organic matter in rhizosphere of pots grown wheat plants, and a further 25% increase was observed both in pot and field grown plants, when tryptophan was added (Hassan & Bano, 2015). Inoculation of *Bacillus cereus* and *P. moraviensis* also increase P contents 25%, K contents 35%, Ca and Mg 20% in rhizosphere soil (Hassan & Bano, 2015).

Soil properties including total bacteria, organic matter and total N showed positive and significant correlation with phosphate solubilizing bacteria (PSB) population (Rfaki et al., 2018). The application of the P Solubilizing Purple Non-sulfur Bacteria biofertilizers improved some soil chemical properties, such as pH_{H20} , EC, NH_4^+ , and available P (Huu et al., 2022). The PSB were able to reduce the soil acidity, meaning that biofertilizers play key roles in raising pH as well; in this case, leading to an increase in NH_4^+ and soluble P concentrations in soil and increase in high available nutrients such as N, P, K, Ca, and Mg (Huu et al., 2022). The increase of total N in soil up to 10% and available P up to 28% following PSB inoculation was reported (Fatima et al., 2022).

CONCLUSION

Phosphate-solubilizing microorganisms play a critical role in maintaining soil health, P cycle, and inorganic P fertilizer efficiency by solubilizing phosphorus in the soil, making it more available to plants. PSM solubilize phosphate by producing metabolites such as organic acids, inorganic acids, H_2S , exopolysaccharide, and siderophore. The use of PSM can help reduce environmental contamination caused by excessive use of chemical fertilizers. PSM can enhance soil organic P cycle through organic P mineralization and decomposition. PSM also improve soil health by maintaining soil properties. It is suggested that PSM shall be utilized in sustainable food crop production.

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